

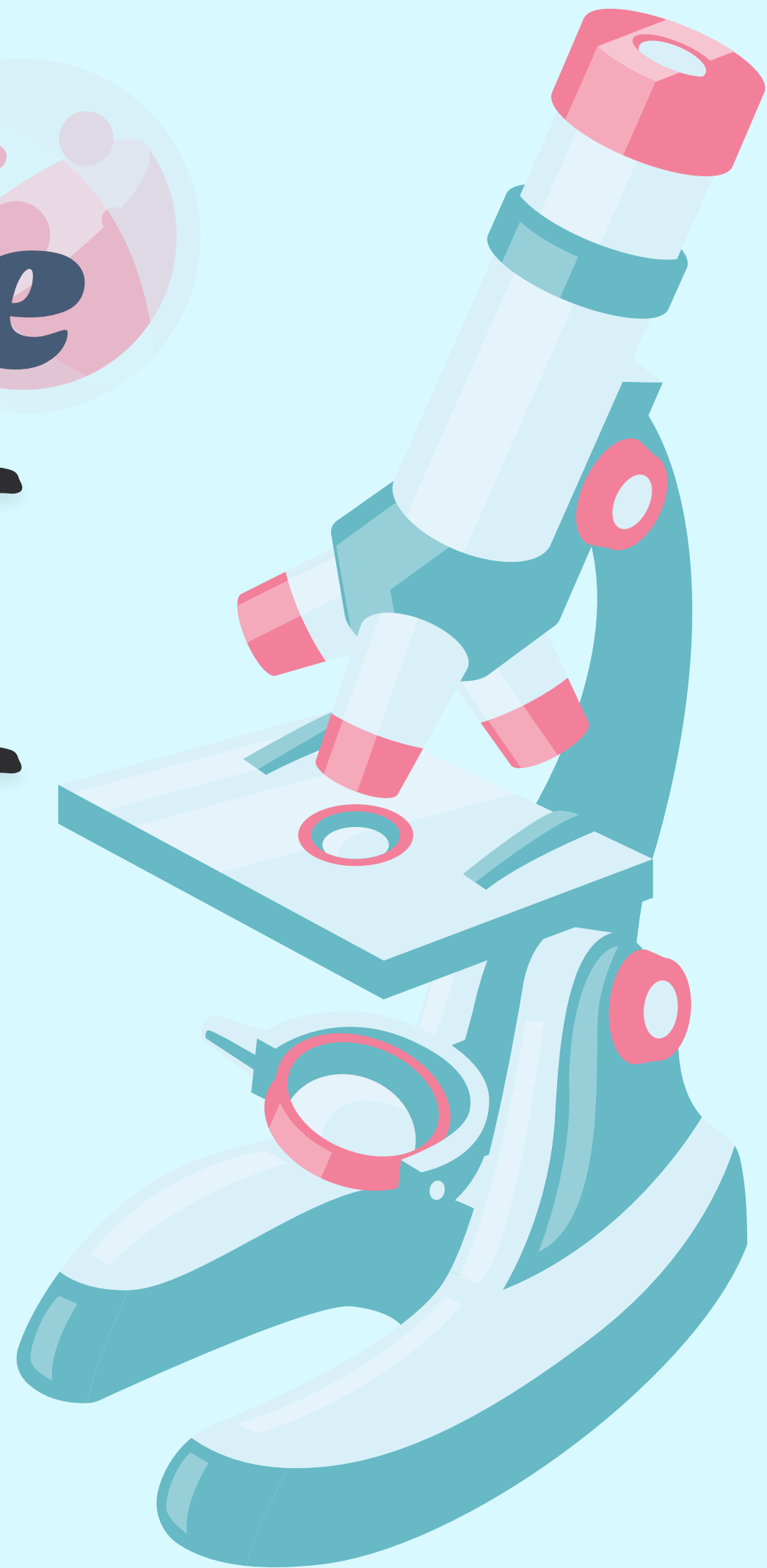
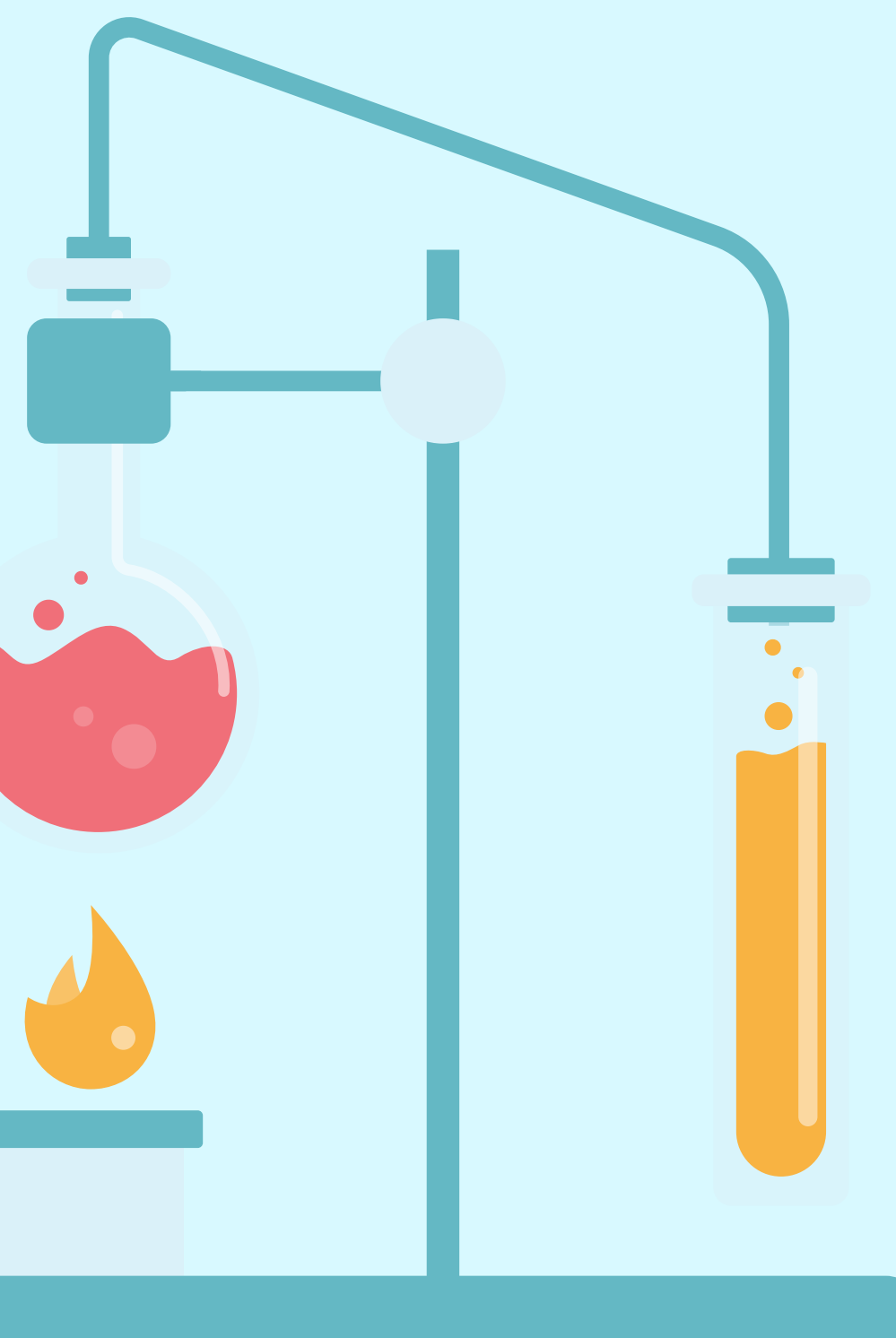


i due per tre

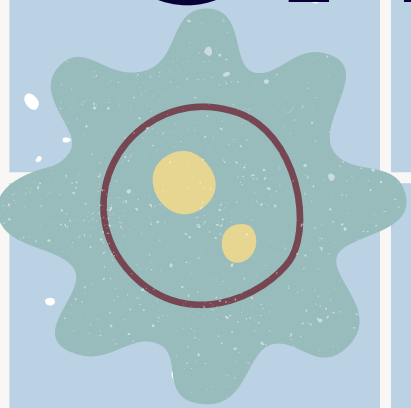
IL LEGAME COVALENTE

2^A

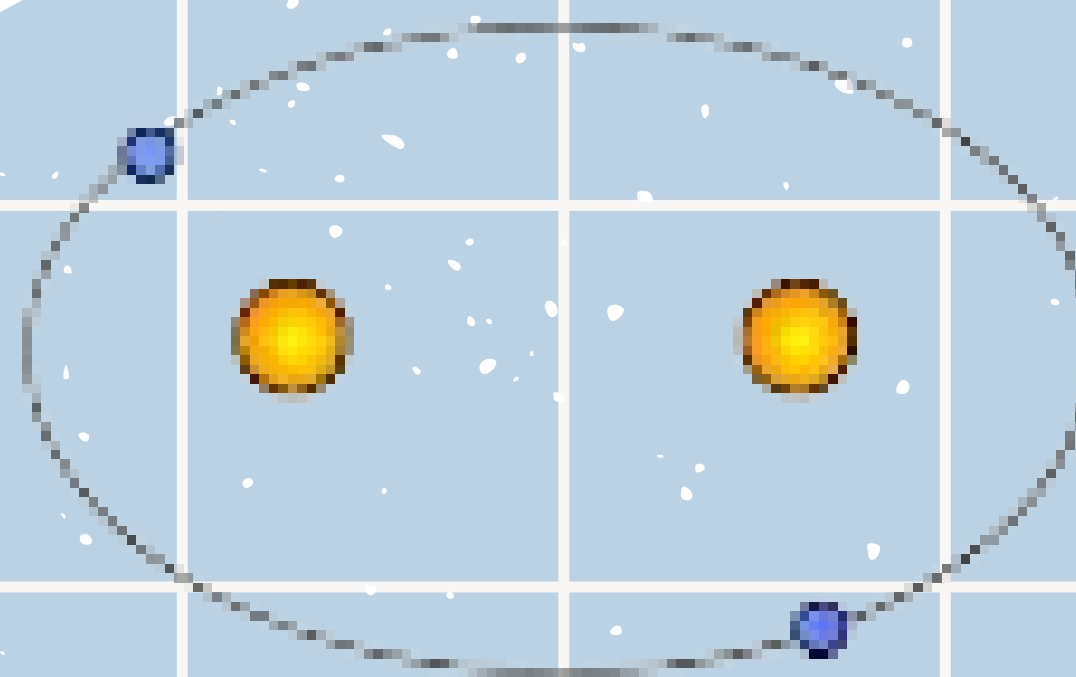
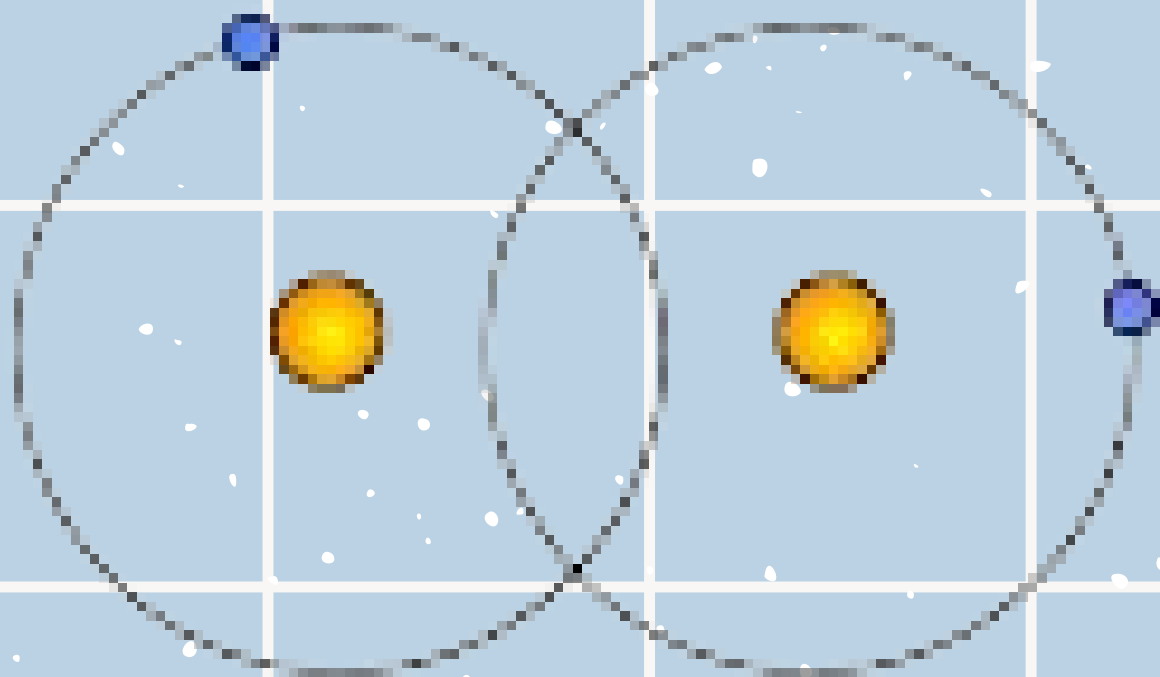
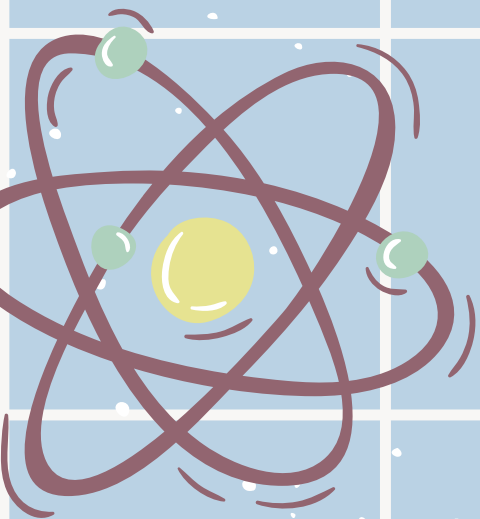
**Scuola Secondaria di
primo grado**



Che cos'è?

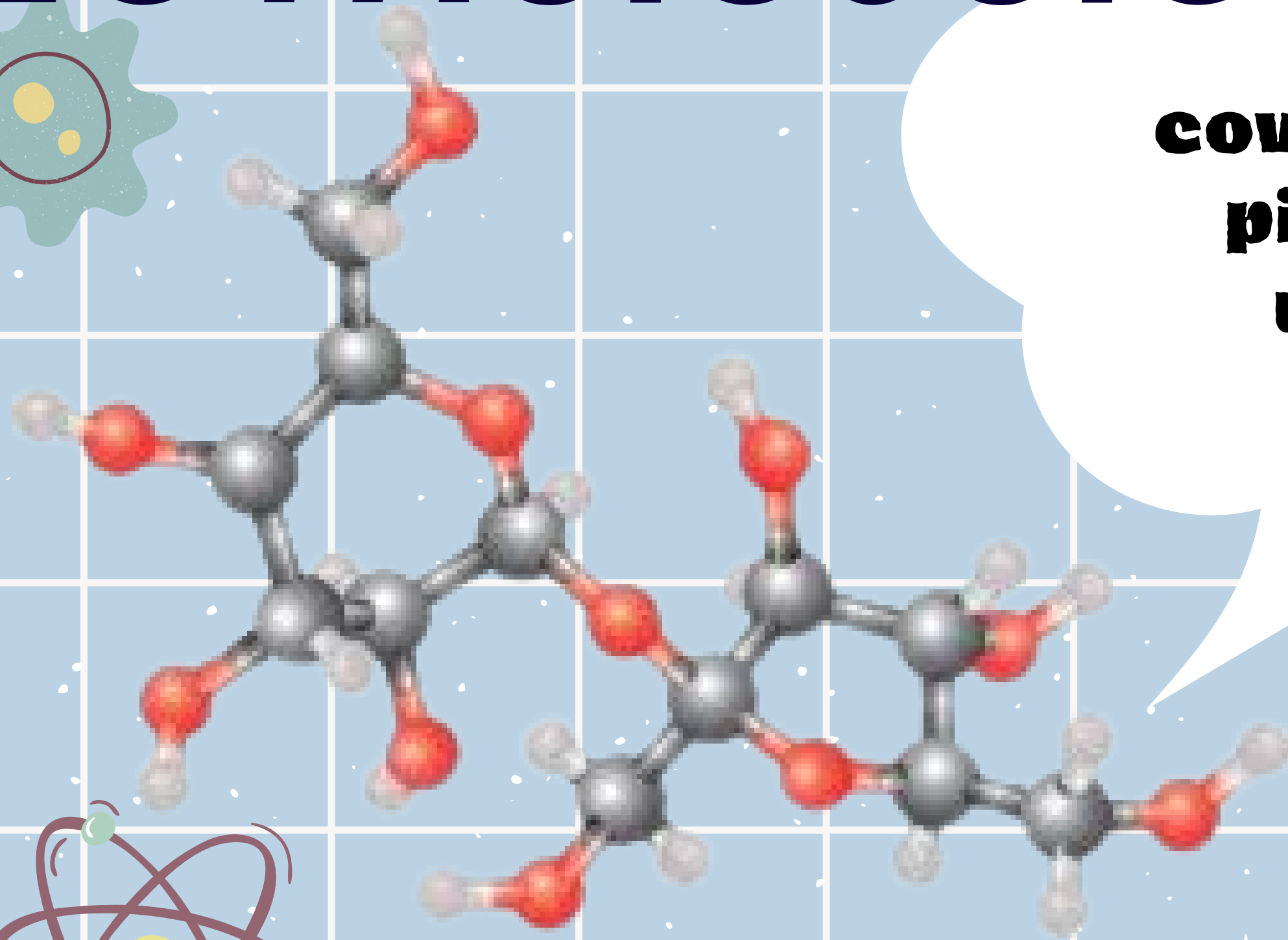
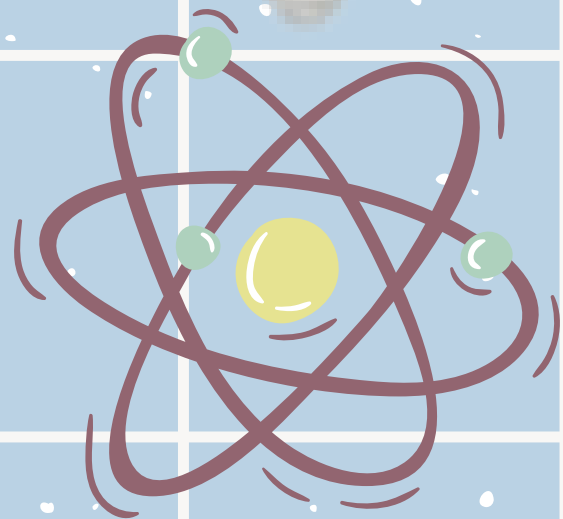
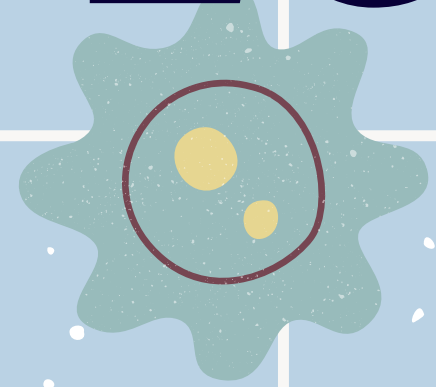


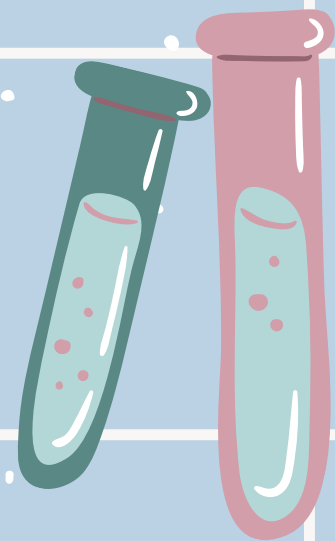
**Si ha un legame
covalente, quando
due o più atomi sono
uniti mediante la
condivisione di due o
più elettroni**



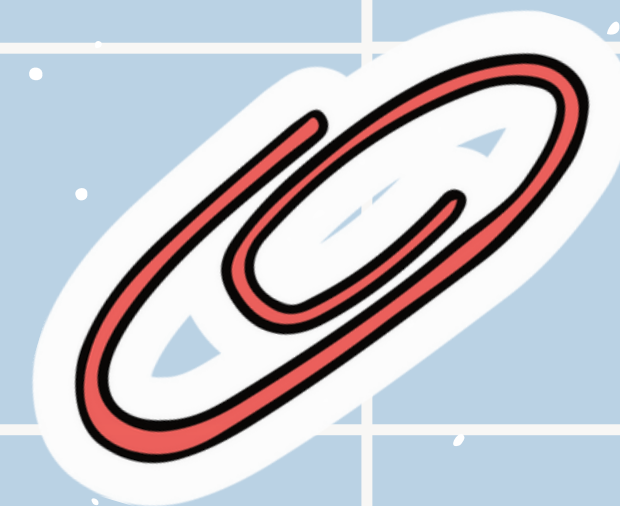
Le molecole

**il legame
covalente tra due o
più atomi forma
una molecola**





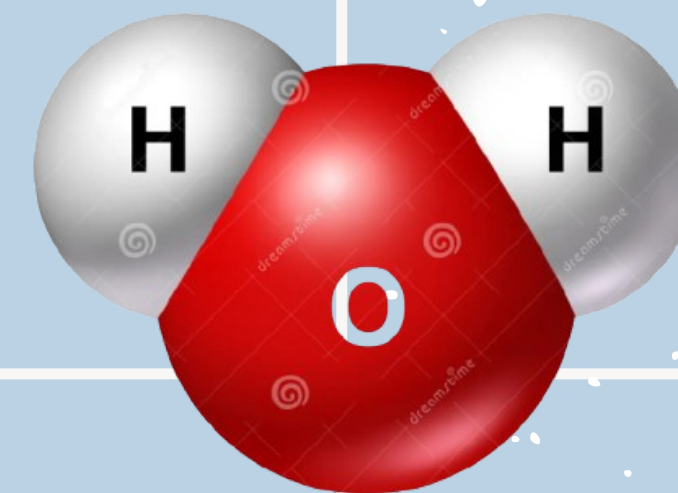
Il legame covalente in natura



molti composti diffusi in natura sono formati grazie al legame covalente:

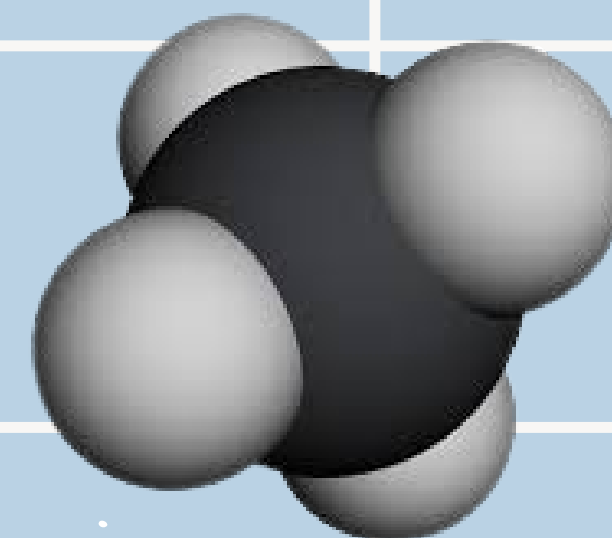
1

molecola d'acqua (H_2O)



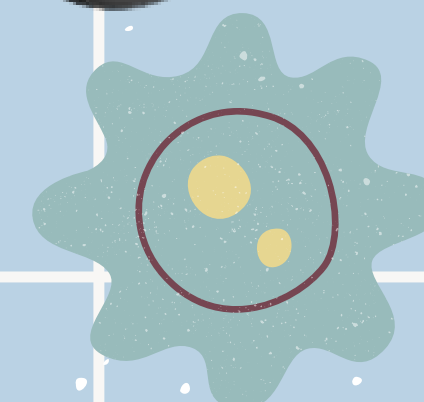
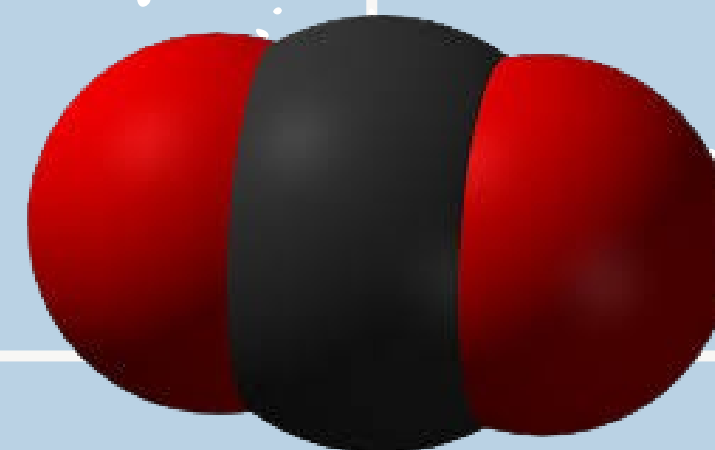
2

molecola di metano (CH_4)



3

**molecola di
anidride carbonica
(CO_2)**





GRAZIE PER L'ATTENZIONE!

Componenti del gruppo:

**Cipullo Alessia, D'Amato Zaira, Della Monica Sole,
Leo Alessia Gerarda, Ruggia Rosario, Viglione Angelo**